

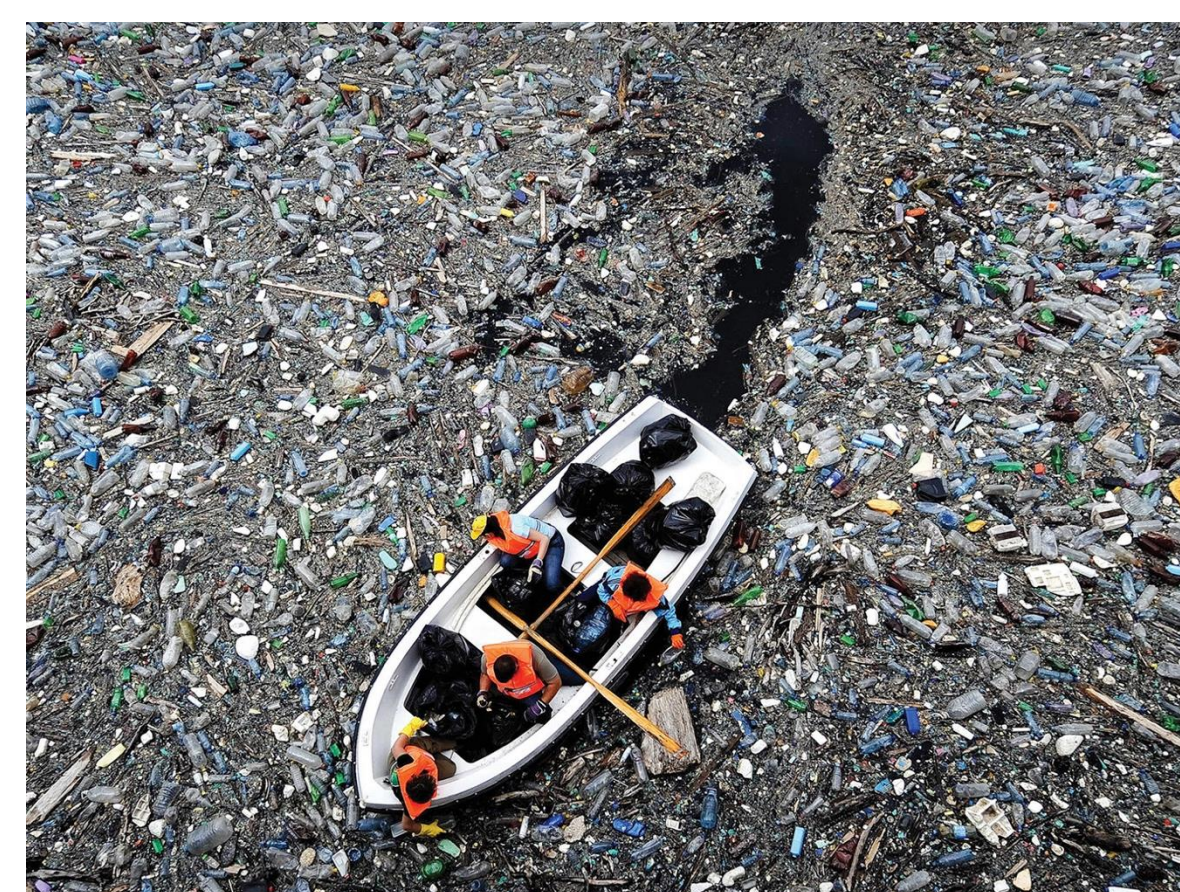
Photoinitiated RAFT Copolymerization of α -Lipoic Acid and Acrylates for Degradable Polymers

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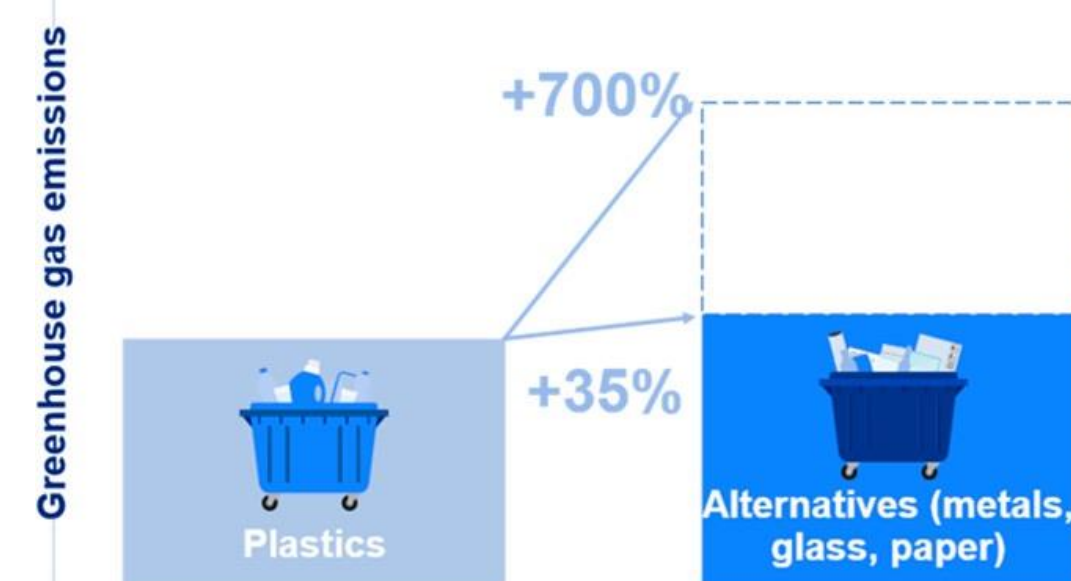
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Introduction



Discarded plastic bottles and garbage on the Vacha River in Bulgaria.

Replacing plastics will **INCREASE GHG EMISSIONS** in most cases

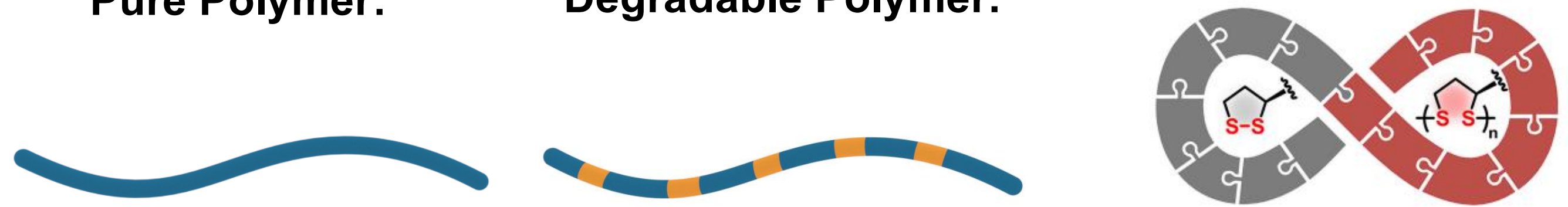


Paper bags weigh significantly more than HDPE bags, leading to higher GHG emissions for production and transportation.

Pure Polymer:

Degradable Polymer:

Closed-loop recyclable poly(disulfides)

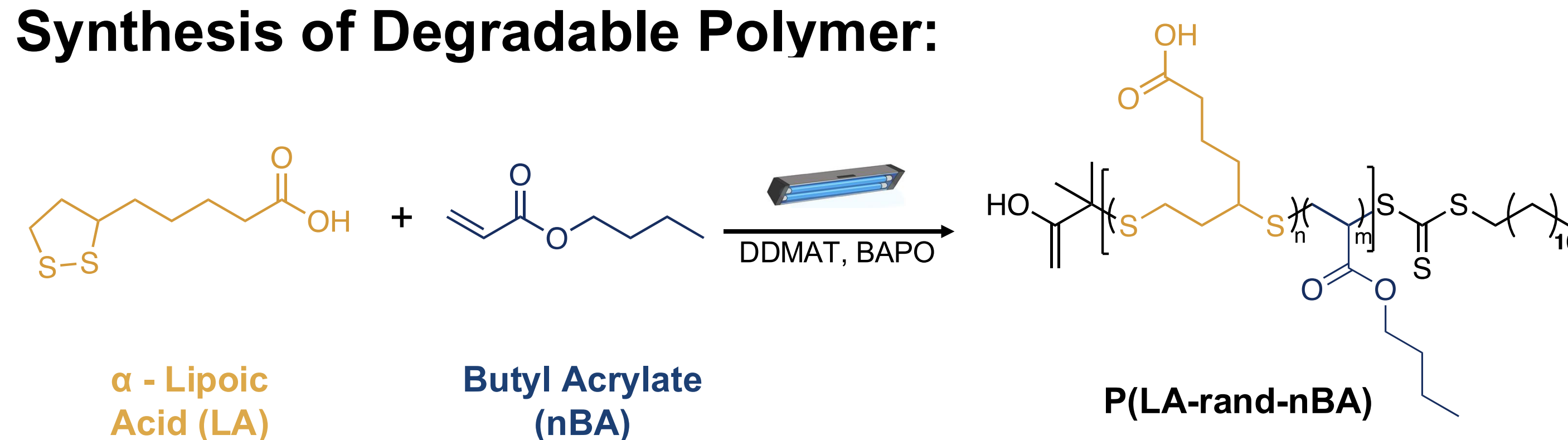


While C-C bonds are difficult to break, disulfide bonds (S-S) are much easier to cleave, making them ideal for forming the degradable backbone of polymers.

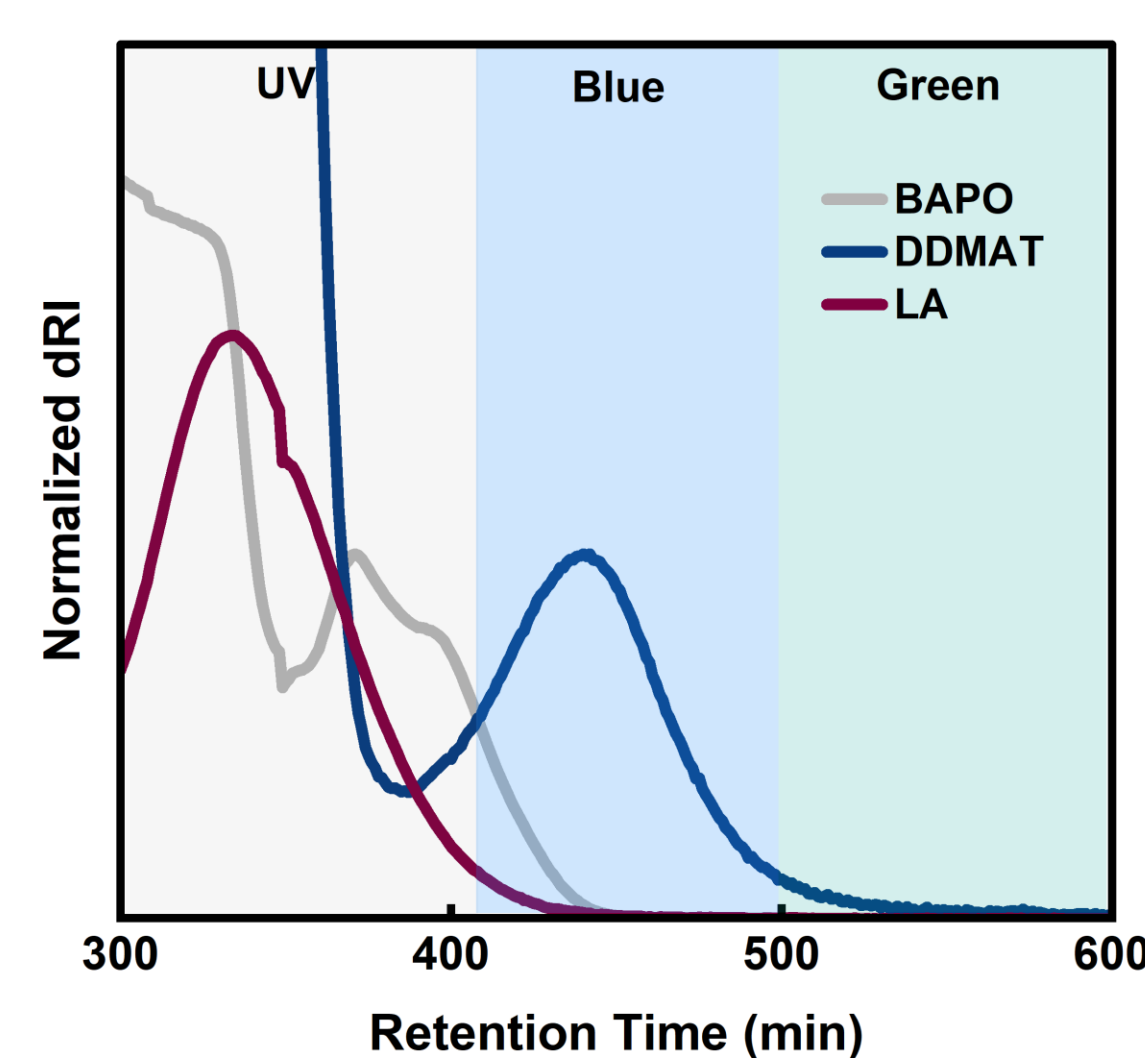
Objective: To develop sustainable polymer, we copolymerized α -Lipoic Acid with acrylate-based monomer through photoinitiated reversible addition fragmentation chain transfer.

Polymerization Mechanism

Synthesis of Degradable Polymer:

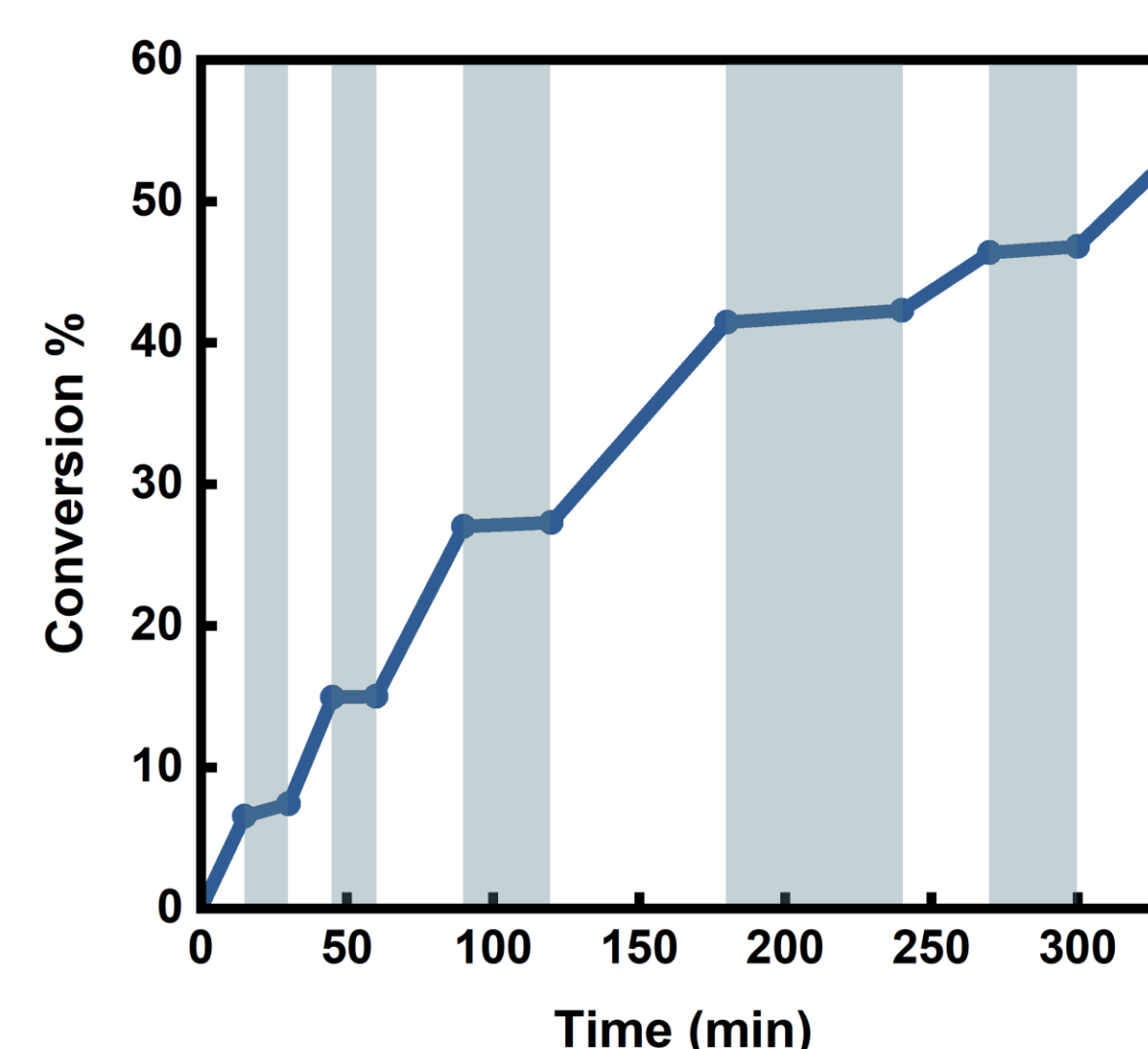


Ultraviolet-Visible Spectroscopy



LA and DDMAT both absorb blue light and generate radicals.

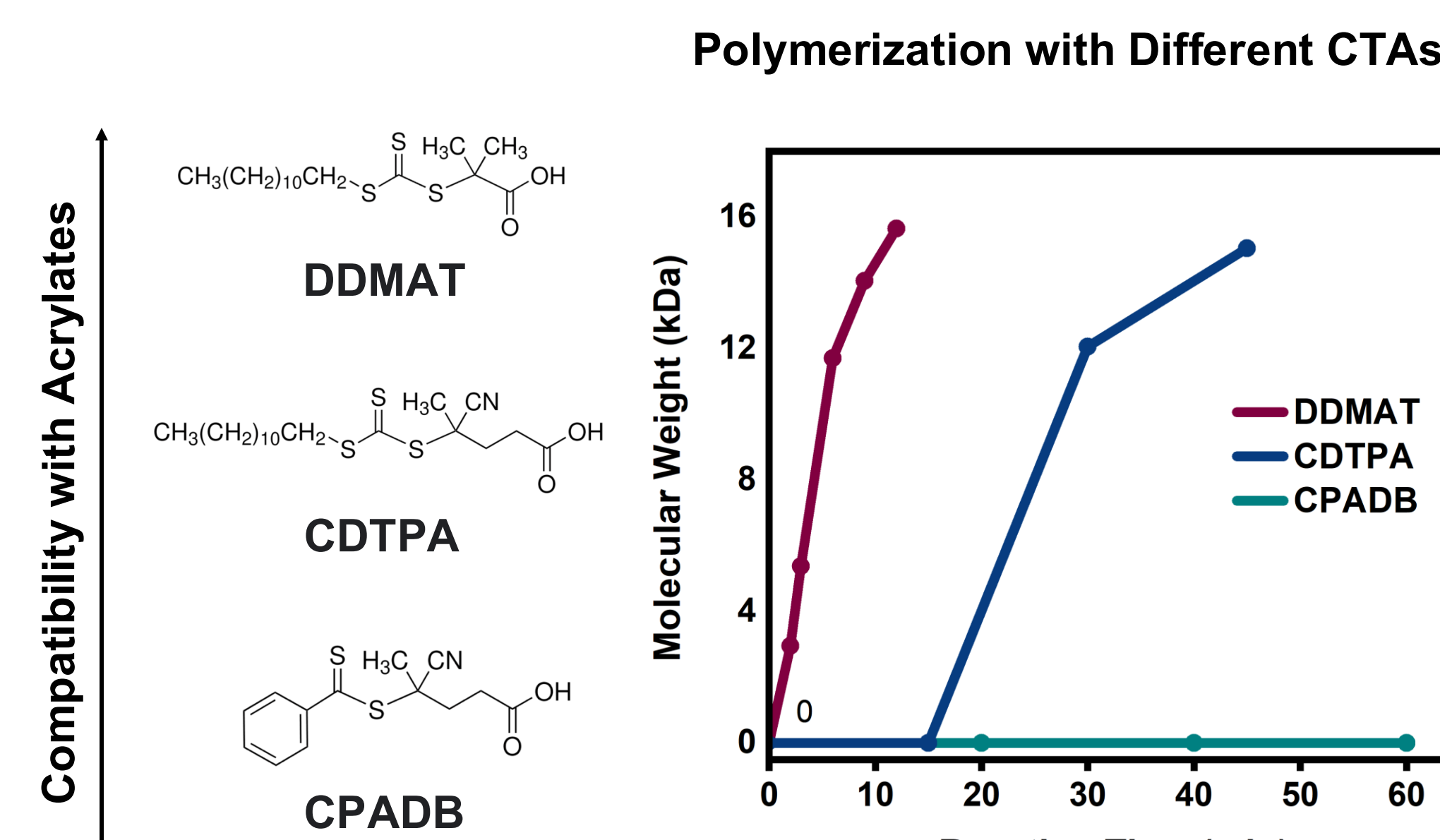
Light Temporal Control



The polymerization only proceed under blue light exposure.

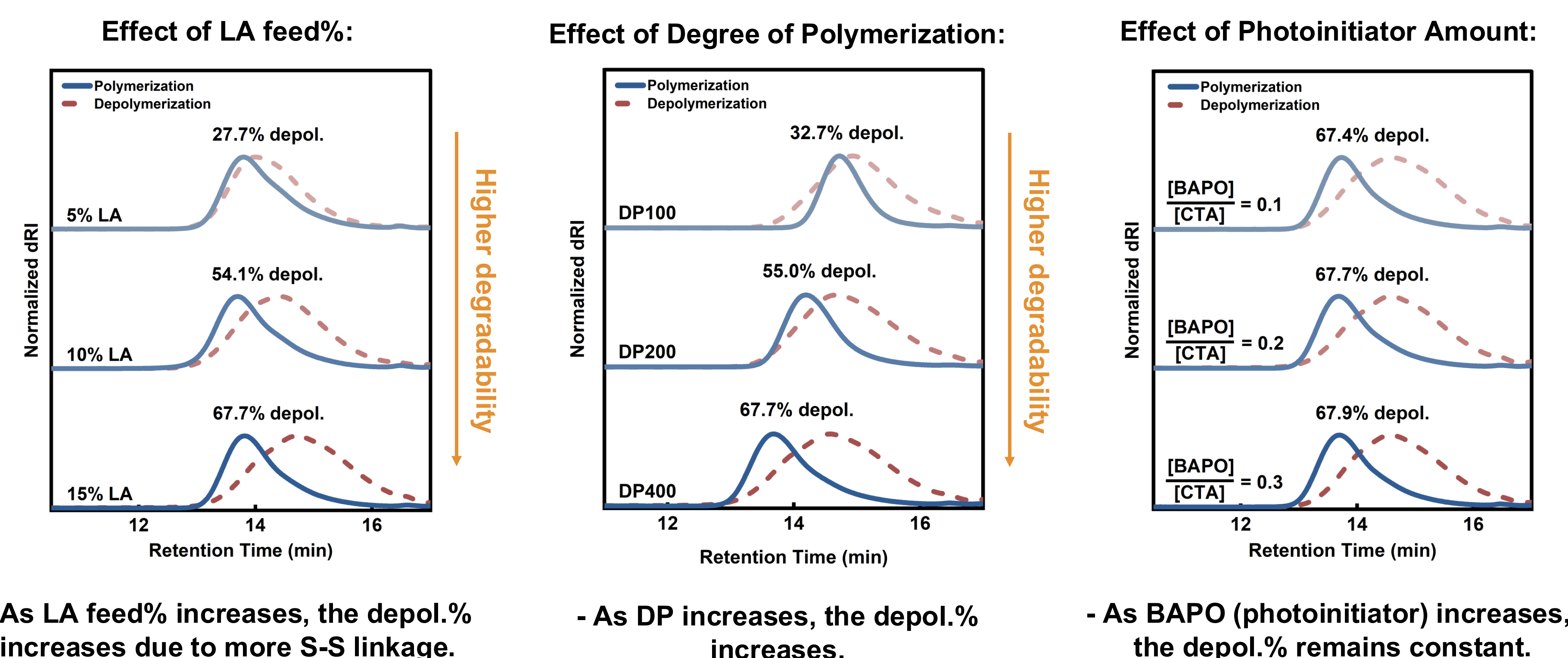
Results and Discussion

Effect of Chain Transfer Agent (CTA):



DDMAT outperforms CDTPA with higher compatibility of nBA and shorter induction time, while CPADB polymerizes none.

Effect of Reaction Conditions:

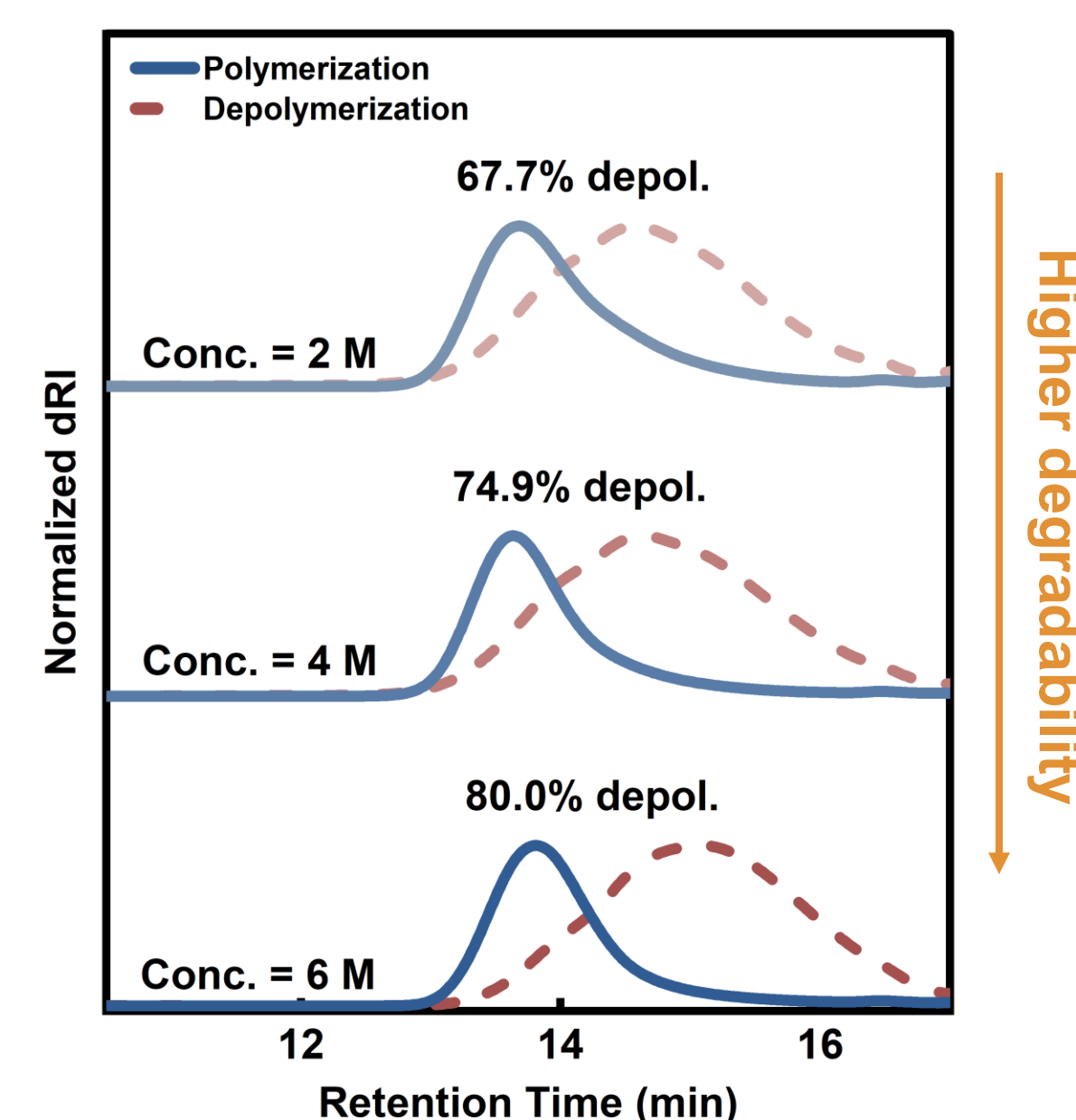


- As LA feed% increases, the depol.% increases due to more S-S linkage.

- As DP increases, the depol.% increases.

- As BAPO (photoinitiator) increases, the depol.% remains constant.

Effect of Concentration:



Conc. = 2 M

More disulfide S-S block in polymer.

Conc. = 6 M

